# Development and Validation of Q-Absorbance Ratio UV Spectrophotometric Method for Simultaneous Determination of Bisoprolol Fumarate and Cilnidipine in Marketed Formulation Besicor C5

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#### Abstract

Introduction: Cardiovascular disorders involve simultaneous administration of more than one drug; hence, numerous combinations, including calcium channel blockers and beta-blockers, are available in the market and are approved by various regulating authorities. Cilnidipine (CLD), a calcium-channel blocker, and Bisoprolol fumarate, a  $\beta$ 1-adrenergic receptor blocking agent are such combinations available on the market. This combination is used in the treatment of hypertension and angina pectoris. Therefore, these drugs must be accurately analyzed in various samples, including laboratory mixtures and marketed formulations. Methods: Simultaneous estimation of Bisoprolol fumarate and CLD in marketed formulation Basicore C5 was carried out using the UV spectrophotometric method in ethanol. The Q-Absorbance ratio method was selected for estimation. The wavelengths used were 224.5 nm, λmax of Bisoprolol fumarate, and 232 nm, an iso-absorptive point of both the drugs. **Results:** Beer's law was obeyed in the concentration range between 2 and 10  $\mu$ g/mL for both the drugs. Recovery studies have validated the analysis results as per ICH guidelines. Accuracy was found between 99.3–99.75% and 99.48–100.06% for Bisoprolol fumarate and CLD, respectively. The limit of detection was 0.0943 and  $0.0751 \,\mu$ g/mL for Bisoprolol fumarate, while 0.0355 and  $0.1141 \,\mu$ g/mL for CLD at 224.5 and 232 nm, respectively. The limit of quantifications was 0.2860 and 0.8983 µg/mL for Bisoprolol fumarate, while 0.1610 and 0.3457 µg/mL for CLD at 224.5 and 232 nm, respectively. Conclusion: The method was simple, reproducible, rapid, and precise. Hence, it could analyze laboratory samples and marketed formulations containing these two drugs.

Key words: Bisoprolol fumarate, Cilnidipine, Q-Absorbance ratio method, Simultaneous estimation, UV Spectrophotometric, Validation

## INTRODUCTION

B isoprolol fumarate (BF) is a competitive antagonist at the  $\beta$ 1-adrenergic receptor site. It is chemically designed as (RS)-1-{4-[(2-isopropoxy ethoxy) methyl] phenoxy}-3-(isopropyl amino) propan-2-ol.<sup>[1,2]</sup>. Cilnidipine (CLD) is a calcium channel blocker with vasodilating effect, used as an antihypertensive agent. It is chemically 1,4-Dihydro-2, 6-dimethyl-4-(3-nitrophenyl)-3, 5-pyridinedicarboxylic acid 2-methoxyethyl (2E)-3-phenyl-2-propenyl ester.<sup>[1,3]</sup> The chemical structures of BF and CLD are shown in Figures 1 and 2, respectively. BF and CLD are available in the market as a fixed-dose combination of 5 and 10 mg, respectively, and this combination has utility in various cardiovascular diseases such as hypertension, angina pectoris, and myocardial infarction.<sup>[4,5]</sup> One of the commercially available combinations is the Besicor C5 tablet by Ajanta Pharma Ltd., India, launched in the Indian

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market in 2017. The present study envisages developing and validating the Q-absorbance UV-spectrophotometric method for simultaneously estimating BF and CLD in their tablet formulation. Determination of BF alone and in combinations with other drugs is reported in the literature by various methods, that is, UV Spectrophotometry,[6-9] HPTLC,<sup>[10-12]</sup> HPLC,<sup>[13-19]</sup> and capillary electrophoresis.<sup>[20-22]</sup> Furthermore, determination of CLD alone or with other drugs by HPTLC,<sup>[23-28]</sup> HPLC,<sup>[29-34]</sup> and spectrophotometry<sup>[30,35-45]</sup> is reported. The literature survey reveals only two reported spectrophotometric methods for simultaneously determining BF and CLDusly.<sup>[45,46]</sup> One economical and well-used analysis method is UV spectrophotometry, which does not require special instrumentation or tedious separation steps as in chromatographic methods. This UV spectrophotometric method further has several reported methods for simultaneous estimation of drugs, out of which the Q-absorbance method is commonly used and further reported in the literature. A literature survey revealed no Q-absorbance ratio method (Q-method) reported for simultaneous analysis of BF and CLD in pharmaceutical preparations. Hence, developing and validating a new Q-method is required to estimate these drugs in a multi-component mixture simultaneously. The present research reports a simple, precise, and accurate Q-method to determine these two drugs simultaneously. Validation method was done as per the ICH guidelines.

#### MATERIALS AND METHODS

#### **Chemicals and reagents**

BF and CLD active pharmaceutical ingredients were supplied as gift samples by Glenmark Pharmaceuticals Ltd, R&D, Taloja, Mumbai, and Alkem Laboratories Ltd, R&D, Mumbai, India, respectively. The certified purity of BF and CLD was 99.05% and 98.97%, respectively. Solvents



Figure 1: Structure of BF



Figure 2: Structure of CLD

used were of analytical grade and purchased from Merck (Germany). The solutions used to prepare standard and test solutions were prepared with ethanol.

Besicor C5 (BF 5 mg and CLD 10 mg) tablets manufactured by Ajanta Pharma Ltd.India were procured from the chemist shop. The solvents were selected based on the solubility of pure drugs and formulation in various solvents such as water, methanol, ethanol, isopropyl alcohol, acetone, and chloroform. The drugs were freely soluble in ethanol. Therefore, ethanol was chosen as a solvent for this study.

#### Instrument and software

Shimadzu PharmaSpec UV-1700 spectrophotometer (Tokyo, Japan), with preset slit width, quartz cuvette of 1 cm was used for UV-visible measurements of the absorption spectra. Regression and statistical analyses were performed with Microsoft Excel 2010.

#### Preparation of standard stock solutions

Graduated long neck flask used to prepare the stock solutions, working solutions, and formulation solutions were calibrated before use. 100 mg of BF and CLD were accurately weighed to prepare standard solutions into a 100 mL volumetric flask by dissolving them in ethanol. Solutions were further diluted with the same solvent to obtain substock solution of 100  $\mu$ g/mL concentration.

## Determination of wavelength of maximum absorbance ( $\lambda$ max) and iso-absorptive point

Solutions of  $10 \ \mu\text{g/mL}$  for both drugs were prepared from a working substock solution, scanned in the range of 200–400 nm against ethanol as blank, and presented in Figures 3 and 4. The overlaying spectrum was also obtained to determine the iso-absorptive point and is presented in Figure 5.



Figure 3: Spectra of different concentrations of BF

## Preparation of working solutions from standard substock solution

The working solutions were prepared by diluting standard substock solutions of BF and CLD in different ratios within their concentration linearity ranges. The solutions were carefully diluted to obtain  $2-10 \,\mu$ g/mL concentration for both to construct the calibration curves.

#### Calibration curve (Linearity)

Accurately measured working substock solution of BF (0.2, 0.4, 0.6, 0.8, and 1.0 mL) and working stock solution of CLD (0.2, 0.4, 0.6, 0.8, and 1.0 mL) was transferred to two different series of 10 mL volumetric flask and diluted up to the mark with ethanol. The absorbance of both solutions at their respective  $\lambda$ max and the iso-absorptive point was collected. The calibration curves were drawn with Microsoft Excel by plotting concentration versus absorbance; an average of three determinations was selected as absorbance.

#### **Procedure - Q method**

The Q-method involves the absorbance ratio at the isoabsorptive point and  $\lambda$ max of any drug. Iso-absorptive point is the wavelength of equal absorptivity of the two species, where the absorbance ratio at a particular wavelength is a constant irrespective of concentration or path length. At an iso-absorptive point, drug solutions of the same concentration exhibit similar absorbance.<sup>[47]</sup>

Determination of spectral characteristics of BF and CLD was performed by scanning UV absorbance spectra of each component individually and together with their mixture over a range between 200 and 400 nm against ethanol as a blank [Figure 5]. The absorbance data for solutions were collected from overlay spectra of two drugs; the two wavelengths designated for the progress of the





Q-absorbance ratio were 224.5 ( $\lambda$ max of BF) and 232 nm (iso-absorptive point). The absorbance of both drugs was determined, and the absorptivity values were calculated at the two selected wavelengths. These values were a mean of three assessments.

The absorptivity coefficients (Qx and Qy) for BF and CLD were determined. The concentration Cx and Cy of both drugs was calculated using the following equation, respectively;

$$Cx = \frac{Qm - Qy}{Qx - Qy} \times \frac{A1}{ax1}$$

$$Cy = \frac{Qm - Qx}{Qy - Qx} \times \frac{A2}{ay1}$$

here, Absorbance ratio Qm = A2/A1, Qx = ax2/ax1and Qy = ay2/ay1

where A1 and A2 are the absorbance's of BF and CLD mixture at 224.5 nm and 232 nm;

ax1 and ax2 are absorptivities of BF at 224.5 nm and 232 nm, respectively, and

ay1 and ay2 are absorptivities of CLD at 224.5 nm and 232 nm, respectively.

#### Application of spectrophotometric methods for marketed formulations besicor C5

Twenty tablets were accurately weighed and powdered in a mortar. Quantity equivalent to one tablet was transferred into a volumetric flask of 100 mL capacity. Powdered drugs were dissolved with vigorous shaking and made up the volume upto 100 mL with ethanol. The solution was filtered through a Whatman filter paper, and residues were rinsed thrice with



Figure 5: UV scan of BF and CLD showing wavelength of maximum absorbance and iso-absorptive points

ethanol. It was diluted appropriately to obtain the desired concentration and scanned over a range between 200 and 400 nm against ethanol as a blank. The absorbance data for the solution were collected.

#### Analytical method validation

The developed UV spectrophotometric method was validated according to ICH Q2 (R1) guidelines. The parameters such as linearity, accuracy, precision, limit of detection (LOD), and limit of quantification (LOQ) were selected for this method validation.

#### Linearity

Linear regression analysis was performed to assess method linearity. The average of three authentic calibration curves, at 2, 4, 6, 8, and 10  $\mu$ g/mL concentrations, was used for the same. The linear regression analysis was performed on the selected wavelengths.

#### Precision and accuracy

Method precision was assessed on two levels intermediate precision and repeatability. The method was tested for these two precision levels based on six determinations of 4 and 8  $\mu$ g/mL for BF and CLD, respectively, at various time intervals on the same day (intra-day precision) and three dissimilar days (inter-day precision). The standard deviation and %Relative Standard Deviation for the inter-day and intra-day precision were determined. Intermediate precision was performed intra-day and inter-day at 100% of test concentration.

#### Accuracy (% recovery)

Accuracy was performed using the standard addition method at three levels of  $100\pm20\%$  test concentration, and the percentage recovery from the spiked solution was determined.

#### LOD and LOQ

As per ICH guidelines, the standard deviation approach determines the LOD and LOQ.LOD and LOQ show the sensitivity of the analytical method, and calculations were done according to the following equation:

$$LOD = 3.3 \frac{\sigma}{m}$$
$$LOQ = 10 \frac{\sigma}{m}$$

Where  $\sigma$  represents the standard deviation of the y-intercepts of the regression line, and mis the mean slope of the calibration curves.

#### **RESULTS AND DISCUSSION**

#### **Method validation**

The Q-absorbance method was validated for linearity, accuracy, precision, and LOD and LOQ of analytes in the experimental conditions, and ICH guidelines were followed. The optical regression characteristic is tabulated in Table 1.

#### Linearity

For constructing calibration curves [Figures 3 and 4], a series of five solutions consisting of 2-10 µg/mL of BF and CLD were analyzed in triplicate. All the experiments maintained optimized conditions, and the regression parameters are summarized in Table 2. Three authentic calibration curves obtained the linear regression equations for the drugs at 224.5 nm [Figure 6] and 232 nm [Figure 7]. Obtained correlation coefficient represents the excellent quality of the calibration curve, as the lower the dispersion of the set of points, the lower the uncertainty of the estimated regression coefficient. As per Beer's Law, linearity was checked and attained in the range of  $2-10 \,\mu\text{g/mL}$ ; selection of the dilution range was based on the ratio of analytes present in the pharmaceutical preparation and the projected concentration in the in vitro dissolution studies. The intercepts of the curves were also not significant, representing the excellent linearity in this range of concentrations.



Figure 6: Calibration curve of BF and CLD at 224.5 nm



Figure 7: Calibration curve of BF and CLD at 232 nm

#### Accuracy and precision

Accuracy was proved by the standard addition method and is represented as percent recovery. The proposed method was accurate as percent recoveries were in the range of 99.3 and 99.75% for BF and 99.48 and 100.06% for CLD [Table 3]. Intra-day, inter-day accuracy, and precision were checked using 4 and 8  $\mu$ g/mL for BF and CLD, respectively. All the analyses were performed in triplicate. For intraday, all the three sets of solutions were analyzed on the same day; for inter-day, all three sets of solutions were analyzed for three consecutive days. Repeatability and intermediate precision were evaluated, and percent relative standard deviation (%RSD) values were calculated. The values of %RSD were found to be less than two, indicating good precision. The precision data are represented in Tables 3 and 4.

#### LOD and LOQ

The concentration levels of LOD and LOQ were determined by analyzing prepared standard solutions of BF and CLD. These LOD and LOQ were found to be in the concentration range of the solution prepared. The LOD were 0.0943 and 0.0751  $\mu$ g/mL for BF, while 0.0355 and 0.1141  $\mu$ g/mL for CLD at 224.5 and 232 nm, respectively. The LOQ was 0.2860 and 0.8983  $\mu$ g/mL for BF, while 0.1610 and 0.3457  $\mu$ g/mL

Table 1: Optical regression characteristic and validation parameters								
Drug		BF	CLD					
Wavelength (nm)	224.5	232 (Iso-absorptive point)	224.5	232 (Iso-absorptive point)				
Linearity range (µg/mL)	2–10	2–10	2–10	2–10				
Regression equation	y=0.0636x+0.0027	y=0.039x+0.0027	y=0.0533x-0.0019	y=0.059x+0.007				
SD of Y-intercept (n=3)	0.0018	0.0012	0.0006	0.0020				
Mean of slope (n=3)	0.0629	0.0534	0.0387	0.0579				
Correlation coefficient (r <sup>2</sup> )	0.9967	0.9984	0.9986	0.9995				
LOD (µg/mL)	0.0943	0.0751	0.0355	0.1141				
LOQ (µg/mL)	0.2860	0.8983	0.1610	0.3457				

BF: Bisoprolol Fumarate, CLD: Cilnidipine

Table 2: Calibration points of the standard curve with SD and %RSD									
Concentration (µg/mL)	ration At 224.5 nm					At 232 nm (Iso-absorptive point)			
	BF CLD				BF	BF		CLD	
	Mean Absorbance SD ( <i>n</i> =3)	%RSD	Mean Absorbance SD ( <i>n</i> =3)	%RSD	Mean Absorbance SD ( <i>n</i> =3)	%RSD	Mean Absorbance SD ( <i>n</i> =3)	%RSD	
2	0.1310±0.002	1.5267	0.0880±0.0015	1.7292	$0.0986 \pm 0.0015$	1.5485	0.1270±0.0020	1.5748	
4	0.2413±0.0020	0.8625	0.1500±0.0015	1.0138	0.2053±0.0020	1.0137	0.2440±0.0030	1.2295	
6	0.4033±0.0020	0.5161	0.2420±0.001	2.0408	0.3226±0.0015	0.4734	0.3616±0.0025	0.6958	
8	0.5176±0.0032	0.6209	0.3150±0.001	1.5625	0.4333±0.0020	0.4803	0.4793±0.0020	0.4342	
10	0.6216±0.0025	0.4048	0.3933±0.001	1.3333	0.5190±0.0030	0.5780	0.5886±0.0035	0.5965	

BF: Bisoprolol Fumarate, CLD: Cilnidipine

	Tab	le 3: Accuracy and	precision data	by % recovery	studies of BF and CLI	C
Drug	Levels %	Known Conc. taken (μg/mL)	Amount added (μg/mL)	Amount found (μg/mL)	Amount of std drug recovered (μg/mL)	% Recovery±SD ( <i>n</i> =3)
BF	80	4	3.2	7.2	7.15	99.3012±0.0070
	100		4	8	7.98	99.7535±0.0141
	120		4.8	8.8	8.76	99.5447±0.0092
CLD	80	8	6.4	14.4	14.35	99.6512±0.0183
	100		8	16	16.01	100.0617±0.0254
	120		9.6	17.6	17.51	99.4843±0.0465

BF: Bisoprolol Fumarate, CLD: Cilnidipine

for CLD at 224.5 and 232 nm, respectively. The calculated LOD and LOQ are also represented in Table 1.

#### Selectivity

The developed method was checked for selectivity by analyzing two different mixture solutions prepared in the laboratory at different ratios; the selections of the concentrations for both the drugs were taken in the linearity range but are above and below the concentration of the marketed formulation, that is, 50% and 150%. Evaluations are done in triplicates. The selectivity results were 98.85–100.95% for BF and 98.57–99.86% for CLD and are presented in Table 5; the results confirm the selectivity of the developed analytical procedures to analyze the various ratios of both drugs. Results of selectivity also rationalize that the method may be utilized to analyze intermediate dissolution

samples of marketed formulation, which will be concentrated below the final marketed solution and can be applied for other developing formulations in other ratios. The results suggest that the method could quantitatively analyze both drugs with high accuracy, selectivity, and precision.

#### Determination of BF and CLD in the formulation

The marketed formulation Besicor C5 (Ajanta Pharma ltd), available in dose ratios of 1:2 (5 mg BF and 10 mg CLD), was analyzed using two different concentrations in the same ratio. The results of the same are presented in Table 6. The standard addition method evaluated the percent recoveries at three different concentrations, 50%, 100%, and 150%. The additions were made to an earlier evaluated tablet solution comprising 4 and 8  $\mu$ g/mL equivalent of BF and CLD. The assay results [Table 7] of

Table 4: Intermediate precision study							
Precision	% Assessment of BF±SD ( <i>n</i> =3)	%RSD	% Estimation of CLD±SD ( <i>n</i> =3)	%RSD			
Intraday Precision	99.6999±0.463	0.4260	99.4614±0.6456	0.7546			
Intermediate Precision	99.7987±0.277	0.5810	99.5535±0.2380	0.6874			

BF: Bisoprolol Fumarate, CLD: Cilnidipine

Table 5: Selectivity study								
Drug	Levels %	Concentration (µg/mL)	Amount of std drug recovered (µg/mL)	% Recovery±SD ( <i>n</i> =3)				
BF	50	2	2.0193	100.9515±0.0057				
	100	4	3.9542	98.8552±0.0024				
	150	6	5.9647	99.4139±0.0006				
CLD	50	4	3.9428	98.5794±0.0015				
	100	8	7.9568	99.4606±0.0009				
	150	12	11.9168	99.2506±0.0007				

BF: Bisoprolol Fumarate, CLD: Cilnidipine

Table 6: Analysis of pharmaceutical dosage form								
Tablet formulation Label claim (mg) An				Amount found		% Assay±SD ( <i>n</i> =3)		
Besicor C5	BF	CLD	BF	CLD	BF	CLD		
	5	10	4.91	9.93	98.2121±0.0680	99.3795±0.0430		

BF: Bisoprolol Fumarate, CLD: Cilnidipine

	Table 7: Determination of formulation by standard addition method								
Drug	Levels %	Conc. taken (μg/mL)	Amount added (μg/mL)	Amount found (μg/mL)	Amount of std drug recovered (μg/mL)	% Recovery±SD ( <i>n</i> =3)			
BF	50		2	6	5.96	99.3349±0.0634			
	100	4	4	8	7.89	98.6251±0.0054			
	150		6	10	9.92	99.2251±0.0129			
CLD	50		4	12	11.93	99.4163±0.0072			
	100	8	8	16	16.27	101.6808±0.0521			
	150		12	20	19.68	98.4161±0.0003			

BF: Bisoprolol Fumarate, CLD: Cilnidipine

the Besicor C5 were concurrent with the actual amount of both analytes. The standard ranges of percent recovery demonstrate the noninterference of formulation excipients in the analysis.

## CONCLUSIONS

A reliable Q absorbance ratio method is developed to simultaneously estimate BF and CLD in combined pharmaceutical formulation using UV Spectrophotometry. The proposed method was precise, accurate, and reproducible and had acceptable recovery of the analytes. The excipients and additives commonly present in the combined dosage form showed no interference in drug analysis. Hence, the method can be suitably adapted for regular quality control analysis of the BF and CLD in a mixture or combined pharmaceutical formulation.

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## WORK SHOULD ATTRIBUTE TO

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